pH measurement of phosphate buffer solutions 1999 - 2000

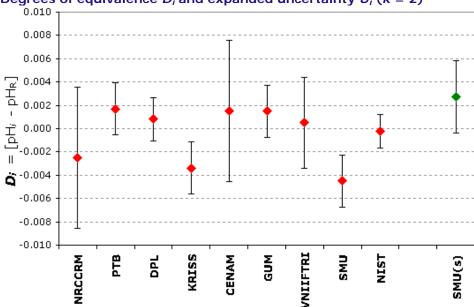
MEASURAND : pH value of phosphate buffer

Sample 1: $[0.025 \text{ mol kg}^{-1} \text{ KH}_2\text{PO}_4 + 0.025 \text{ mol kg}^{-1} \text{ Na}_2\text{HPO}_4]$

Measurements at 15 °C

NOMINAL VALUE: pH = 6.9 at 25 °C

Degrees of equivalence D_i and expanded uncertainty U_i (k = 2)



• indicates the degree of equivalence for SMU resulting from the subsequent bilateral comparison between SMU and PTB.

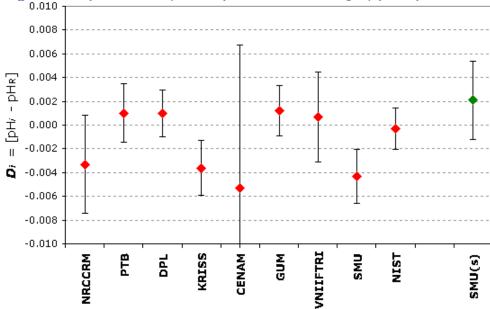
MEASURAND: pH value of phosphate buffer

Sample 1: $[0.025 \text{ mol kg}^{-1} \text{ KH}_2\text{PO}_4 + 0.025 \text{ mol kg}^{-1} \text{ Na}_2\text{HPO}_4]$

Measurements at 25 °C

NOMINAL VALUE: pH = 6.9 at 25 °C

Degrees of equivalence D_i and expanded uncertainty U_i (k = 2)



• indicates the degree of equivalence for SMU resulting from a subsequent bilateral comparison between SMU and PTB.

MEASURAND: pH value of phosphate buffer

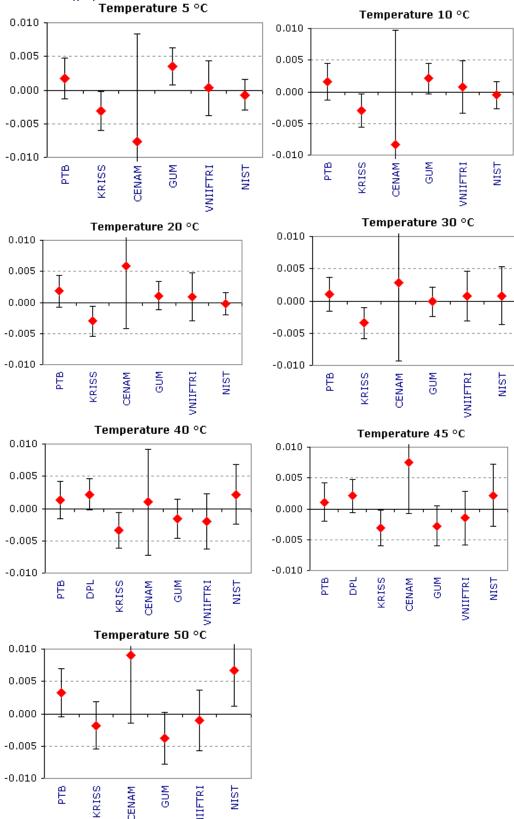
Sample 1: $[0.025 \text{ mol kg}^{-1} \text{ KH}_2\text{PO}_4 + 0.025 \text{ mol kg}^{-1} \text{ Na}_2\text{HPO}_4]$

Measurements at 5 °C, 10 °C, 20 °C, 30 °C, 40 °C, 45 °C and 50 °C

NOMINAL VALUE: pH = 6.9 at 25 °C

Degrees of equivalence D_i and expanded uncertainty U_i (k = 2) for each temperature

Click on the graph for a closer view



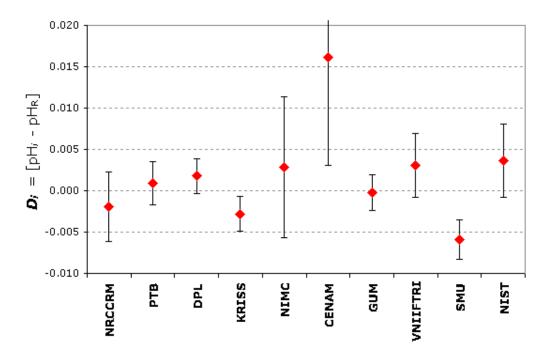
MEASURAND: pH value of phosphate buffer

Sample 2: $[0.02 \text{ mol kg}^{-1} \text{ KH}_2\text{PO}_4 + 0.02 \text{ mol kg}^{-1} \text{ Na}_2\text{HPO}_4]$

Measurements at 25 °C

NOMINAL VALUE : pH = 6.9 at 25 °C

Degrees of equivalence D_i and expanded uncertainties U_i (k = 2)



MEASURAND: pH value of phosphate buffer

Sample 2: $[0.02 \text{ mol kg}^{-1} \text{ KH}_2\text{PO}_4 + 0.02 \text{ mol kg}^{-1} \text{ Na}_2\text{HPO}_4]$

Measurements at 37 °C

NOMINAL VALUE: pH = 6.9 at 25 °C

Degrees of equivalence D_i and expanded uncertainty U_i (k = 2)

